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Transient radiative heat transfer within a suspension of coal particles undergoing steam gasification

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Abstract Transient radiative heat transfer in chemical reacting media is examined for a non-isothermal, non-gray, absorbing, emitting, and Mie-scattering suspension of coal particles, whose radiative properties vary with time as the particles undergo shrinking by endothermic gasification. A numerical model that incorporates parallel filtered collision-based Monte Carlo ray tracing, finite volume method, and explicit Euler time integration scheme is formulated for solving the unsteady energy equation that couples the radiative heat flux with the chemical kinetics. Variation of radiative properties, attenuation characteristics, temperature profiles, and extent of the chemical reaction are reported as a function of time. It is found that radiation in the visible and near IR spectrum incident on a cloud of coal particles greater than $2.5\mu\text{m}$ is more likely to be forward scattered than absorbed, but the opposite is true as the particles shrink below $1.3\mu\text{m}$. The medium becomes optically thinner as the particles shrink and this effect is more pronounced for smaller initial coal particles because these offer higher volume fraction to particle diameter ratio and, consequently, attain higher temperatures, reaction rates, and shrinking rates.

Nomenclature

a Mass specific active surface area ($\text{m}^2\cdot\text{kg}^{-1}$)
 A Active surface (m^2); overall layer absorptance
 c_p Specific heat at constant pressure ($\text{J}\cdot\text{kg}^{-1}\text{K}^{-1}$)

C_p Molar specific heat at constant pressure ($\text{J}\cdot\text{mol}^{-1}\text{K}^{-1}$)
 d Particle diameter (m)
 E_a Activation energy ($\text{J}\cdot\text{mol}^{-1}$)
 E_n Exponential integral function
 f_v Particle volume fraction
 I Intensity of radiation ($\text{W}\cdot\text{m}^{-2}\text{sr}^{-1}$)
 k Imaginary part of the complex refractive index
 k Reaction rate constant ($\text{mol}\cdot\text{m}^{-2}\text{s}^{-1}\text{Pa}^{-1}$, Pa^{-1})
 k_0 Frequency factor ($\text{mol}\cdot\text{m}^{-2}\text{Pa}^{-1}$, Pa^{-1})
 L Layer thickness (m)
 m Mass (kg); complex refractive index
 M Molar mass ($\text{kg}\cdot\text{mol}^{-1}$)
 n Molar amount (mol)
 \dot{n} Molar flow rate ($\text{mol}\cdot\text{s}^{-1}$)
 n Real part of the complex refractive index
 n_{rays} Number of rays in the sample for each Monte Carlo run
 p Pressure (Pa)
 q Power (W)
 q'' Heat flux ($\text{W}\cdot\text{m}^{-2}$)
 Q Efficiency factor
 r Reaction rate ($\text{mol}\cdot\text{kg}^{-1}\text{s}^{-1}$)
 R Universal gas constant = $8.31441\text{J}\cdot\text{mol}^{-1}\text{K}^{-1}$; overall layer reflectance
 S Radiative source function ($\text{W}\cdot\text{m}^{-2}\text{sr}^{-1}$)
 t Time (s)
 T Temperature (K); overall layer transmittance
 x Cartesian coordinate in x -direction (m)
 X Extent of reaction

Greek symbols

δ Error in the temperature profile; relative difference in the reaction extent
 Δt Time step interval (s)
 Δx Spatial mesh size in x -direction; thickness of a sub-layer (m)
 ε Total hemispherical emissivity
 κ Absorption coefficient (m^{-1})
 λ Wavelength (m)

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θ	Polar angle
ρ	Density ($\text{kg}\cdot\text{m}^{-3}$)
σ	Stefan–Boltzmann constant = $5.67 \times 10^{-8} \text{ W}\cdot\text{m}^{-2} \text{ K}^{-4}$
σ_s	Scattering coefficient (m^{-1})
τ	Optical thickness
ξ	Size parameter
Φ	Scattering phase function
ω	Albedo

Subscripts

0	Initial value
a	Absorbed
b	Boundary
b_1	Boundary at $x = 0$
baseline	Baseline
e	Emitted
i	Index of a chemical component
in	Incident
j	Index of a sub-layer
k	Dummy index of summation
l	Index of reaction rate constant
L	Whole layer
P	Planck mean value
r	Radiative
ref	Reference
s	Scattering
source	Source of radiation
surr	Surroundings
t	Transmitted
V	Volume
x	In x -direction
λ	Spectral

Superscripts

n	Time level
*	Dummy variable of integration

Dimensionless groups

Bi	Biot number
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1 Introduction

Radiation heat transfer in participating media undergoing a chemical transformation, where the chemical species and their radiation properties vary with time as the reaction progresses, is a complex transient phenomenon often encountered in combustion, atmospheric, and other chemical processes. Examples of previous pertinent studies dealing with transient radiative heat transfer¹ include radiative cooling of an absorbing-scattering layer containing droplets [23],

radiative heating in semitransparent materials [24], and coupled radiation–conduction heat transfer solved by Monte Carlo (MC) [18], by a semi-analytical approach based on the exponential kernel method [10], and by numerical integration of the radiative flux [15]. Previous modeling studies of transient radiative transfer coupled to chemical kinetics include a two-phase model to predict ignition temperatures in an exothermic catalytic chemical reactor [28], a 2D model of the ignition of polymethyl-methacrylate based on the discrete ordinates method [6], a combined convection–conduction–radiation heat transfer model for an oil-fired furnace using the Hottel’s zone and MC methods [12], and for calcium carbonate packed-bed undergoing thermochemical decomposition using the Rosseland diffusion approximation for the internal radiative transport and MC for the radiative heat flux at boundaries [11].

An emerging application field of transient radiative heat transfer in chemical reacting systems is solar thermochemistry [26]. Solar thermochemical reactors make use of concentrated solar radiation—with radiative power fluxes exceeding 500 Wcm^{-2} —as the source of high-temperature process heat. Examples of solar thermochemical processes include the thermal decomposition of limestone, the thermal reduction of metal oxides, the thermal cracking of natural gas, and the thermal gasification of coal. The direct irradiation of the chemical reactants provides efficient energy transfer to the reaction site, bypassing the limitations imposed by indirect heat transfer through reactor walls. Modeling such directly irradiated solar chemical reactors is crucial for anticipating the consequences of a given design decision on the reactor’s performance [21]. Previous studies of radiative transfer within particle suspensions exposed to concentrated solar radiation include steady-state models based on the discrete ordinates method [3], on the six-flux method [16], and on the MC method [17, 8, 9].

The thermodynamics and chemical aspects of solar coal gasification have been previously examined [30, 20]. This paper deals with the transient radiative heat transfer within a suspension of reacting coal particles that are directly exposed to concentrated solar radiation. It considers a non-isothermal, non-gray, absorbing, emitting, and Mie-scattering suspension of coal particles, whose spectral and directional dependent radiative properties vary with time as the particles undergo shrinking by steam-gasification.

The radiative heat flux is computed using the MC technique. Since MC is based on tracing random paths of a large number of generic rays, the computation time increases significantly with accuracy, especially in transient simulations requiring convergence of iterative algorithms at each time step and MC runs for each iteration. This situation is typically encountered in chemical reacting systems. The MC technique used in the present study makes use of parallelization and the Savitzky–Golay smoothing filter for improving accuracy and shortening computation time. Five time integration

¹The term “transient radiative heat transfer” refers in this work to an unsteady heat transfer problem with time-dependent temperature field but quasi-steady radiative transfer

schemes are tested for marching in time. The accuracy of MC combined with the different time integration schemes is presented in the Appendix. It was determined by solving for the temperature distribution within a plane layer of a gray-isotropic participating medium, and comparing the results with those obtained using a semi-analytical method.

2 Analysis

The system domain is depicted schematically in Fig. 1. It consists of a suspension of coal particles that is directly exposed to an external source of thermal radiation, created by concentrated solar power. The particle suspension is modeled as a non-isothermal, non-gray, absorbing, emitting, and anisotropically scattering layer surrounded by black plane boundaries at 0 K (non-participating surroundings). The coal particles undergo shrinking with time due to their gasification process, as observed experimentally [20]. Each particle is assumed isothermal (a good approximation for $Bi \ll 1$), spherical (a generally good assumption for most irregularly shaped randomly oriented particles), and having independent scattering (as justified by referring to independent/dependent scattering regime map, as given by Tien and Drohler [27] for the range of particle volume fractions and size parameters used in this study). For simplification, the gas phase is taken to be a non-participating medium (its contribution to the energy balance has been estimated to be less than 5% [8]). The particle suspension is assumed monodispersed, a simplifying assumption that allows for a better understanding of the effect of particle size variation with time. Further, for the purpose of better elucidating the physical phenomena involved in the pure interaction of radiation with the chemical reacting particles, convection and conduction are omitted from this analysis. No closed analytical solution is available for this problem.

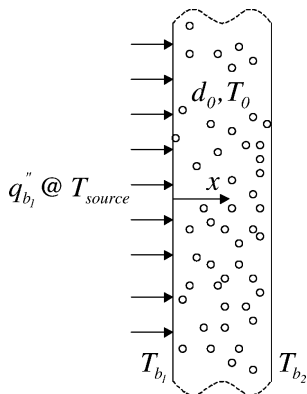
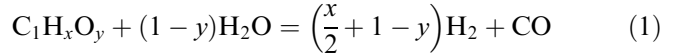


Fig. 1 Scheme of a plane layer of a non-isothermal, non-gray, absorbing-emitting-anisotropically scattering suspension of reacting coal particles (with initial particle diameter d_0), exposed to an external source of concentrated thermal radiation, and undergoing shrinking by steam-gasification

2.1 Chemical kinetics

The steam-gasification of coal is a complex process, but the overall chemical conversion can be represented by the net reaction:



where x and y are the elemental molar ratios of H/C and O/C in coal, respectively, e.g. $x = 0.43$, $y = 0.03$ for anthracite. The kinetic modeling is based on the analysis of elementary reaction mechanisms describing reversible adsorption/desorption processes and irreversible surface chemistry, derived by Müller et al. [20]. Assuming plug flow conditions and pseudo first-order reactions, a set of simplified rate laws of the Langmuir–Hinshelwood type are formulated to describe the formation and consumption of each gas species in terms of rate constants k_l ,

$$r_{H_2O} = -k_1 a p_{H_2O} \quad (2)$$

$$r_{H_2} = k_1 a p_{H_2O} \quad (3)$$

$$r_{CO} = k_1 a p_{H_2O} + 2k_2 a p_{CO_2} - 2k_1 a k_3 p_{H_2O} p_{CO} \quad (4)$$

$$r_{CO_2} = -k_2 a p_{CO_2} + k_1 a k_3 p_{H_2O} p_{CO} \quad (5)$$

where r_i is the reaction rate of gas species i ($i = H_2, H_2O, CO$, and CO_2) for heterogeneous surface reactions,

$$-r_i = \frac{1}{m_c} \frac{dn_i}{dt} = \frac{a}{A} \frac{dn_i}{dt} \quad (\text{mol kg}^{-1} \text{s}^{-1}) \quad (6)$$

where A is the active surface area (including pore surface within the particle), and a the mass specific active surface area, assumed constant. Note that the units of the reaction rates are normalized to the mass of coal, and the units of the rate constants are k_1 ($\text{mol} \cdot \text{m}^{-2} \text{s}^{-1} \text{Pa}^{-1}$), k_2 ($\text{mol} \cdot \text{m}^{-2} \text{s}^{-1} \text{Pa}^{-1}$), and k_3 (Pa^{-1}). The partial pressures are calculated assuming ideal gases,

$$p_i = p_{\text{tot}} \frac{\dot{n}_i}{\sum_k \dot{n}_k} \quad (7)$$

The system of four coupled differential Eqs. 2, 3, 4, and 5 was solved numerically by iterating on the values of k_l , $l = 1, 2, 3$, to minimize the difference between theoretically calculated and experimentally measured molar flow rates of products. The experimental data were collected using a fluidized bed reactor of coal particles in steam directly exposed to concentrated thermal radiation [20]. The temperature dependence of each k_l is determined by imposing an Arrhenius-type rate law:

$$k_l(T) = k_{0,l} \cdot \exp\left(\frac{-E_{A,l}}{RT}\right) \quad (8)$$

The apparent activation energies and corresponding frequency factors are listed in Table 1. Since the k_l

represent complex reaction mechanisms rather than elementary steps, negative values are possible.

2.2 Radiative properties

Mie theory is applied. Absorption and scattering efficiency factors, $Q_{a\lambda}, Q_{s\lambda}$, and the phase function for scattering, Φ_λ , were obtained using the subroutine BHMIE [1] for a given size parameter $\xi = \pi d/\lambda$ and complex refractive index $m_\lambda = n_\lambda + ik_\lambda$. The m_λ was taken for propane soot [2], and its real and imaginary parts were interpolated linearly and are plotted as a function of wavelength in Fig. 2. For a given volume fraction and particle diameter, the absorption and scattering coefficients are found from the corresponding efficiency factors. Note that since the particle diameter changes with time as the reaction progresses, $\kappa_\lambda, \sigma_{s,\lambda}$ and Φ_λ , need to be calculated at each time step for every wavelength and size parameter.

2.3 Mass and energy conservation equations

The rate of decomposition of coal particles is obtained by mass balance on carbon:

$$r_C = -r_{CO} - r_{CO_2} \quad (9)$$

Table 1 Arrhenius kinetic parameters for coal gasification

	E_a	k_0
K_1a	$163.8 \times 10^3 \text{ J}\cdot\text{mol}^{-1}$	$3.7 \times 10^3 \text{ mol}\cdot\text{kg}^{-1} \text{ s}^{-1} \text{ Pa}^{-1}$
K_2a	$-405.5 \times 10^3 \text{ J}\cdot\text{mol}^{-1}$	$2.87 \times 10^{-23} \text{ mol}\cdot\text{kg}^{-1} \text{ s}^{-1} \text{ Pa}^{-1}$
K_3	$-480.7 \times 10^3 \text{ J}\cdot\text{mol}^{-1}$	$2.56 \times 10^{-24} \text{ Pa}^{-1}$

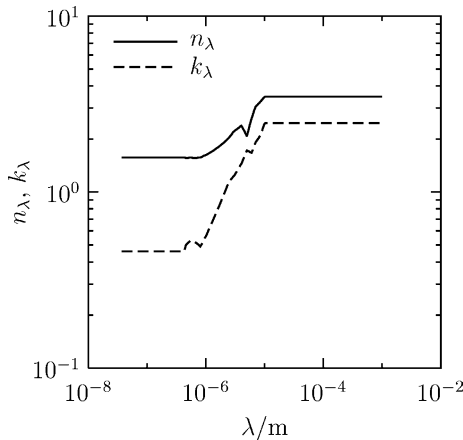


Fig. 2 Spectral distribution of the real and imaginary parts of the complex refractive index of soot, $m_\lambda = n_\lambda + ik_\lambda$ (Dalzell and Sarofin 1969)

Thus, the actual number of C moles at time t is:

$$n_C = n_{0,C} + \int_{t_0}^t m_C r_C dt \quad (10)$$

and the particle diameter at time t is found as

$$d = \left(\frac{6M_C n_C}{\pi \rho_C} \right)^{1/3} \quad (11)$$

Applying energy conservation and accounting for the enthalpy change of C in the solid phase and of H_2, H_2O, CO and CO_2 in the gas phase,

$$\frac{d(n_C h_C)}{dt} = q_r - m_C (r_{H_2} h_{H_2} + r_{H_2O} h_{H_2O} + r_{CO} h_{CO} + r_{CO_2} h_{CO_2}) \quad (12)$$

where q_r is the volume integral of the divergence of the radiative flux, given by [25, 19].

$$q_r = q_a - q_e = - \int_{\Delta x} \frac{\partial q_r''}{\partial x} dx \quad (13)$$

Assuming each particle being isothermal, the following expression for the particle temperature is derived:

$$T = \int_{t_0}^t \left(q_r - m_C \sum_i r_i h_i \right) \frac{dt}{C_{p,C} n_C} + T_0 \quad (14)$$

The initial conditions are:

$$n_{\text{particle}}(t=0) = n_{C,0} = \frac{\rho_C f_{v,C,0} V}{M_C} \quad (15)$$

$$T_{\text{particle}}(t=0) = T_0 \quad (16)$$

The boundary conditions are:

$$\vec{q}_{\text{in},b}''(0,t) = q_{\text{in},b_1}'' \hat{I} \quad (17)$$

$$\vec{q}_{\text{in},b}''(L,t) = 0 \quad (18)$$

The incident collimated radiative flux q_{b_1}'' has the spectral distribution of a blackbody at temperature 5780 K—to simulate concentrated solar power—and is perpendicular to the boundary. Radiation from surroundings is eliminated by setting them black ($\varepsilon = 1$) and cold ($T_{\text{surr}} = 0 \text{ K}$).

3 Numerical solution

The explicit Euler scheme was employed to discretize the time integrals of mass and energy conservation equations. This scheme was chosen after a series of numerical tests with five times integration schemes (explicit and implicit Euler schemes, Crank–Nicolson, 2nd and 4th order Runge-Kutta) because it exhibited stability when

combined with the MC's stochastic errors. Appendix contains the results of this comparative examination. The infinite slab was divided into a number of parallel sub-layers j of volume V_j , each with uniform temperature, composition, and radiative properties. The number of C moles in sub-layer j at time step $n+1$ is, in a discretized form,

$$n_{C,j}^{n+1} = n_{C,j}^n + m_C r_{C,j}^n \Delta t \quad (19)$$

The discretized equation for temperature, Eq. 14, takes the form:

$$T_j^{n+1} = \left(q_{r,j}^n - m_C \sum_i r_{i,j}^n h_{i,j}^n \right) \frac{\Delta t}{C_{p,C,j}^n n_{C,j}^n} + T_j^n \quad (20)$$

The net radiative power to each elemental volume V_j is calculated at every time step n . The emitted term is

$$q_{e,j} = 4V_j \kappa_p \sigma T_j^4 \quad (21)$$

where κ_p is the Planck mean absorption coefficient, $\kappa_p = (1/\sigma T^4) \int_0^\infty \kappa_\lambda e_{\lambda,b}(\lambda, T) d\lambda$. The absorbed term q_a is calculated by forward collision-based MC [29, 4] at every time step n because the radiative properties vary with time as the particle diameter decreases. The MPI parallelization [14, 13] and low pass Savitzky–Golay smoothing filter [22] are incorporated to reduce computational time and improve accuracy of the solution (see Appendix). Parallelization by ray is applied for ray tracing and by spatial region for the radiative properties and energy balance.

3.1 Results

Baseline parameters are listed in Table 2, and are used for each run unless stated otherwise. The total reaction time simulated was 5 s. Figure 3 shows the temperatures obtained. The variation of the temperature of the first

sub-layer—which is directly exposed to the incoming flux—with time is shown in Fig. 3a for three values of the initial particle diameter: 2.5, 5, and 7.5 μm . For all three cases, this first sub-layer is rapidly heated and reaches peak temperature levels of 1910, 1826, and 1802 K within 0.005, 0.01, and 0.012 s, respectively. Afterwards, the temperature decreases slightly as a result of the endothermic reaction and of the decrease in the net absorbed radiative power. As it will be shown in the analysis that follows, the latter effect is caused by the decrease in the absorption coefficient, which in turn depends on the ratio of volume fraction to particle diameter. The temperature distribution across the slab after various reaction times is shown in Fig. 3b. The baseline parameters were used for an initial particle size of 5 μm . While after 0.003 s the highest temperature was obtained in the 1st sub-layer, the peak switched to the 6th sub-layer after 0.009 s. The temperature profile becomes more uniform with time as the optical thickness of the medium decreases.

The extent of the chemical conversion is defined as $X = 1 - n_C/n_{C,0}$. The X for the whole suspension is plotted in Fig. 4a as a function of time for various initial particle diameters, while the local X is plotted in Fig. 4b as a function of location for various reaction times. The fastest conversion occurs for $d_0 = 2.5 \mu\text{m}$ because of the corresponding higher temperatures (as seen in Fig. 3a). After 5 s, the overall reaction extent reached 83, 72, and 69% for the three initial particle diameters 2.5, 5, and 7.5 μm , respectively. Interestingly, Fig. 4b indicates that the local reaction extent is not maximum at the 1st sub-layer but at the 6th sub-layer ($z/L = 0.183$), where, for $d_0 = 5 \mu\text{m}$, reaches 77% after 5 s.

The radiative properties of the suspension are shown in Figs. 5, 6, 7, 8, and 9. Figure 5 shows the absorption and scattering efficiency factors as a function of the particle diameter for $\lambda = 0.5$ and 1.8 μm , which correspond to the peaks of blackbody spectral emissive power at 5780 (solar radiation) and 1600 K (typical reactor

Table 2 Baseline parameters used for the coal particle suspension

Parameter	Value	Unit
$C_{p,C}$	$0.11 + 38.94 \times 10^{-3} \cdot T - 0.15 \times 10^6 \cdot T^{-2} - 17.38 \times 10^{-6} \cdot T^2$	$\text{J} \cdot \text{mol}^{-1} \text{K}^{-1}$
$C_{p,\text{CO}}$	$30.96 + 2.44 \times 10^{-3} \cdot T - 0.28 \times 10^6 \cdot T^{-2}$	$\text{J} \cdot \text{mol}^{-1} \text{K}^{-1}$
C_{p,CO_2}	$51.13 + 4.37 \times 10^{-3} \cdot T - 1.47 \times 10^6 \cdot T^{-2}$	$\text{J} \cdot \text{mol}^{-1} \text{K}^{-1}$
C_{p,H_2}	$26.88 + 3.59 \times 10^{-3} \cdot T + 0.11 \times 10^6 \cdot T^{-2}$	$\text{J} \cdot \text{mol}^{-1} \text{K}^{-1}$
$C_{p,\text{H}_2\text{O}}$	$50.42 + 4.18 \times 10^{-3} \cdot T - 0.85 \times 10^6 \cdot T^{-2}$	$\text{J} \cdot \text{mol}^{-1} \text{K}^{-1}$
D_0	5×10^{-6}	m
$f_{v,0,C}$	1×10^{-5}	—
L	0.1	m
$L/\Delta x$	30	—
n_{rays}	2×10^6	—
q''_{b1}	10^6	$\text{W} \cdot \text{m}^{-2}$
q''_{b2}	0	$\text{W} \cdot \text{m}^{-2}$
Δt	3×10^{-4} for $d_0 > 2.5 \mu\text{m}$ 1×10^{-5} for $d_0 = 2.5 \mu\text{m}$	s
T_0	298.15	K
T_{surr}	0	K
T_{source}	5780	K
ε_b	1	—
ρ_C	815	$\text{kg} \cdot \text{m}^{-3}$

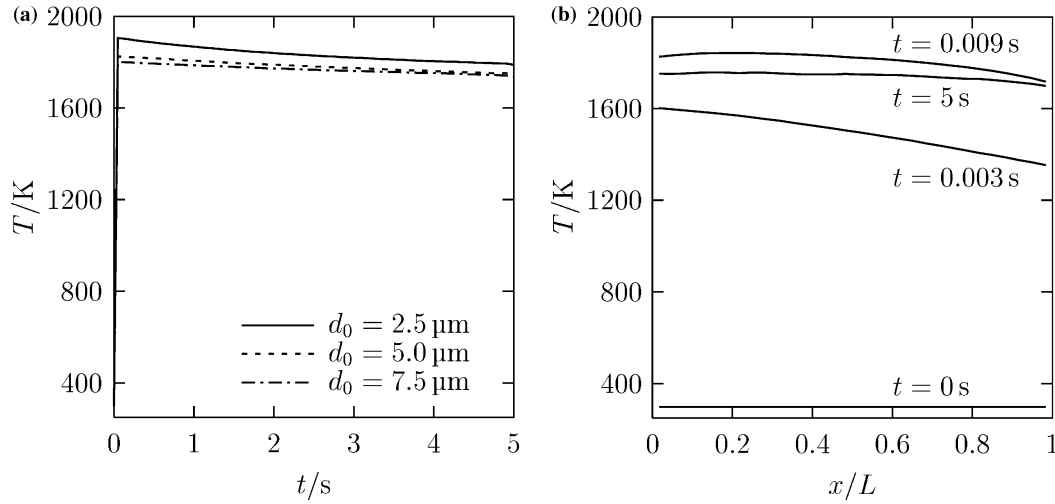
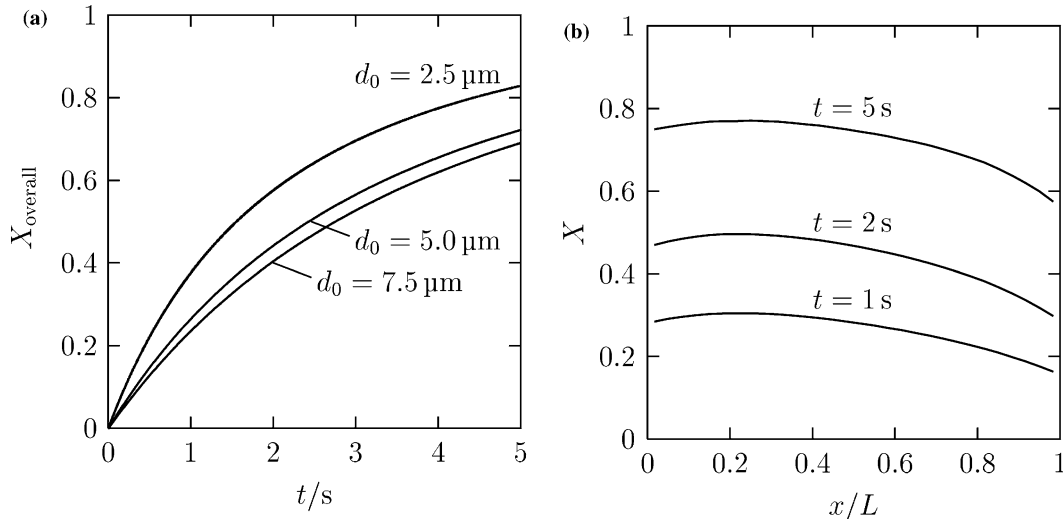


Fig. 3 Variation of the temperature for a suspension of coal particles undergoing steam-gasification. **a** Time-variation of temperature for the first sub-layer—exposed directly to the external radiative flux—for particles of initially $d_0 = 2.5, 5$, and $7.5 \mu m$. **b** Temperature profile across the layer for particles of initially $d_0 = 5 \mu m$, at $t = 0, 0.003, 0.009$, and $5 s$

wall temperature), respectively. For particle sizes in the range 10^{-6} – 10^{-5} m, $Q_{a\lambda}$ decreases with particle diameter while $Q_{s\lambda}$ remains constant. The curves cross each other, implying that $\Omega_\lambda > 0.5$ for $d_0 > 1.3$ and 2.5×10^{-6} m at $\lambda = 0.5$ and $1.8 \mu m$, respectively. Thus, radiation in the visible and near IR spectrum (shown only for $\lambda = 0.5$ and $1.8 \mu m$) incident on a cloud of particles greater than 2.5×10^{-6} m is more likely to be scattered than absorbed, but the opposite is true as the particles shrink below 1.3×10^{-6} m. Both efficiencies are higher at $\lambda = 1.8 \mu m$ than at $\lambda = 0.5 \mu m$, and the medium becomes optically thicker for longer wavelengths. The variation

of the absorption and scattering coefficients with time is shown for the first sub-layer in Fig. 6a and b, respectively. Curves are plotted for two radiation wavelength $\lambda = 0.5$ and $1.8 \mu m$, and for three initial particle diameters, $d_0 = 2.5, 5$, and $7.5 \mu m$. Both coefficients are higher for smaller initial particle diameters and decrease monotonically with time as the particles shrink due exclusively to the decrease in the ratio of volume fraction to particle diameter, because, as seen in Fig. 5, the corresponding efficiencies increase or remain approximately constant with shrinking particles. As far as the effect of the spectrum is concerned, both coefficients increase with longer wavelength (shown only for $\lambda = 0.5$ and $1.8 \mu m$), consistent with the results of Fig. 5. The scattering phase function Φ_λ as a function of the polar angle θ measured from the direction of incident ray (Φ_λ is independent of the azimuthal angle for spherical particles) is shown in Fig. 7 for the 1st sub-layer at $t = 0$ and $5 s$. Curves are plotted for two radiation wavelength: $\lambda = 0.5$ (Fig. 7a) and $1.8 \mu m$ (Fig. 7b). At all times, scattering is predominantly in the forward direction and an order of magnitude higher for $0.5 \mu m$ than for $1.8 \mu m$ radiation. Back scattering remains almost

Fig. 4 Variation of the extent of the reaction X for a suspension of coal particles undergoing steam-gasification. **a** X_{overall} as a function of time for particles of initially $d_0 = 2.5, 5$, and $7.5 \mu m$. **b** X profile across the layer for particles of initially $d_0 = 5 \mu m$, at $t = 1, 2$, and $5 s$



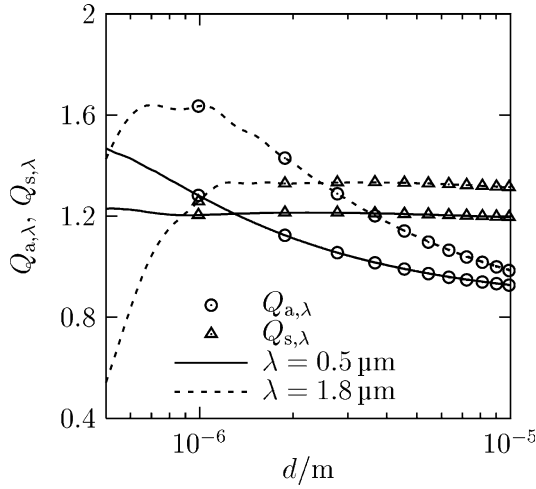


Fig. 5 Radiative properties (absorption and scattering efficiency factors) as a function of particle diameter for radiation at $\lambda = 0.5$ and $1.8 \mu\text{m}$

unchanged with time. The oscillatory character of the phase function can be observed especially at the shorter wavelength. It is to expect that the phase function will oscillate less for poly dispersions due to the superposition of Mie scattering distributions for different particle diameters. From the point of view of energy transfer to a solar chemical reactor, the dominant forward scattering is a desired property because of the deeper penetration of the incident solar radiation and lesser radiation losses.

The transient radiative behavior of the whole slab can in part be examined in Fig. 8, where its spectral optical thickness, defined as $\tau_{\lambda,L} = \int_0^L (\kappa_{\lambda} + \sigma_{s,\lambda}) dx$, is plotted as a function of time at $\lambda = 0.5$ and $1.8 \mu\text{m}$, and for $d_0 = 2.5, 5$, and $7.5 \mu\text{m}$. As expected from the values of absorption and scattering coefficients, the medium

becomes optically thinner as the particles shrink. This effect is more pronounced for smaller initial particles because these offer higher f_v/d and, consequently, reach higher temperatures and gasification rates. As observed already in Fig. 6 for the 1st sub-layer, the whole medium becomes thicker at longer wavelengths.

Finally, the attenuation characteristics of the whole slab are shown in Fig. 9, where the variation of the overall reflectance R , absorptance A , and transmittance T of the slab with time are plotted for three initial particle diameters, $d_0 = 2.5, 5$, and $7.5 \mu\text{m}$. Obviously $R + A + T = 1$. Transmission of incident radiation is dominant. The portion of absorbed radiation decreases logarithmically with time as the particles shrink and the medium becomes optically thinner. For example, for $d_0 = 5 \mu\text{m}$, it amounts initially to 27%, and decreases to 13% after 5 s. The reflectance is more than one order of magnitude smaller. In practice, the reactor's walls absorb, reflect, and re-emit the transmitted incident solar radiation.

The error in the reaction extent caused by time discretization and by the statistical character of the MC method was assessed by varying the time step ($\Delta t = 3$ and 1.5×10^{-4} s), and the number of rays ($n_{\text{rays}} = 2 \times 10^6$ and 10^7). The relative differences in the temperature and reaction extent were calculated by:

$$\delta_T(x, T) = \left| 1 - \frac{T(x, T)}{T_{\text{baseline}}(x, T)} \right| \times 100\% \quad (22)$$

$$\delta_X(x, T) = \left| \frac{X(x, t) - X_{\text{baseline}}(x, t)}{1 - X_{\text{baseline}}(x, t)} \right| \times 100\% \quad (23)$$

where the subscript “baseline” denotes the reference case with baseline parameters. The maximal relative difference in the temperature was $\delta_T = 2.55\%$ when varying Δt and $\delta_T = 0.5\%$ when varying n_{rays} . The maximal relative difference in the reaction extent was $\delta_X = 9.22 \times 10^{-2}\%$ when varying Δt and $\delta_X = 0.7\%$ when varying n_{rays} .

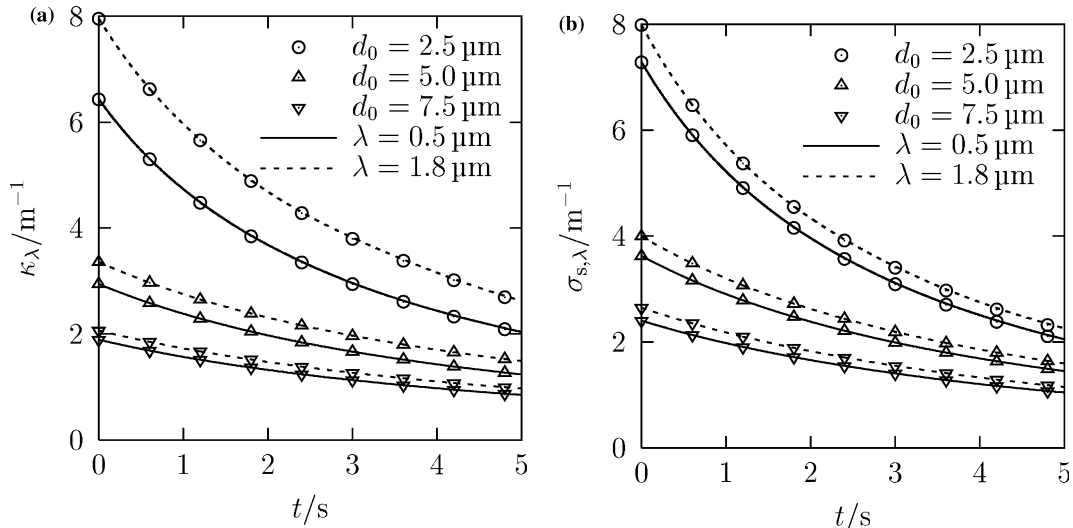


Fig. 6 Variation of the absorption (a) and scattering (b) coefficients with time for the first sub-layer of a suspension of coal particles undergoing steam-gasification, for radiation at $\lambda = 0.5$ and $1.8 \mu\text{m}$, and for initial particle diameters $d_0 = 2.5, 5$, and $7.5 \mu\text{m}$

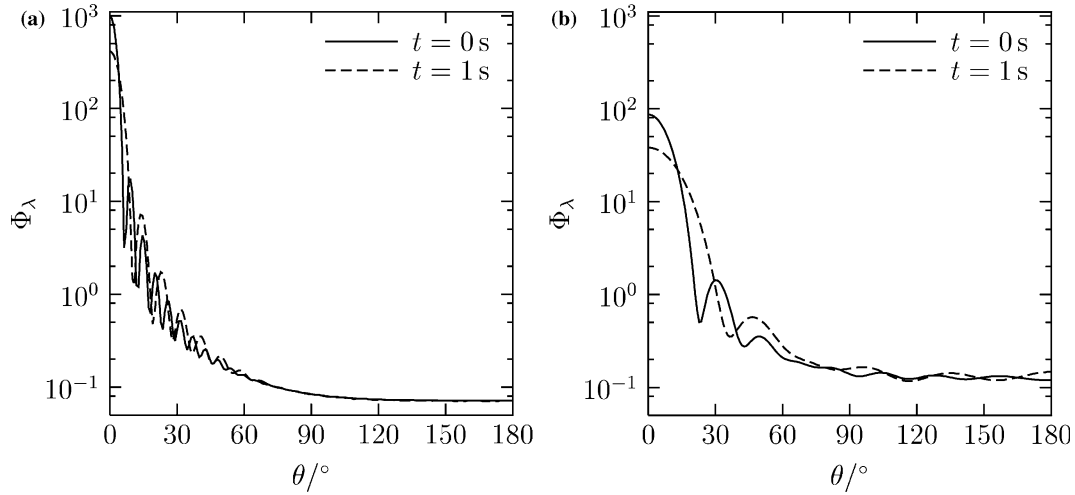


Fig. 7 Scattering phase function for the first sub-layer at $t = 0$ and 5 s, and for radiation at $\lambda = 0.5 \mu\text{m}$ (a) and at $\lambda = 1.8 \mu\text{m}$ (b)

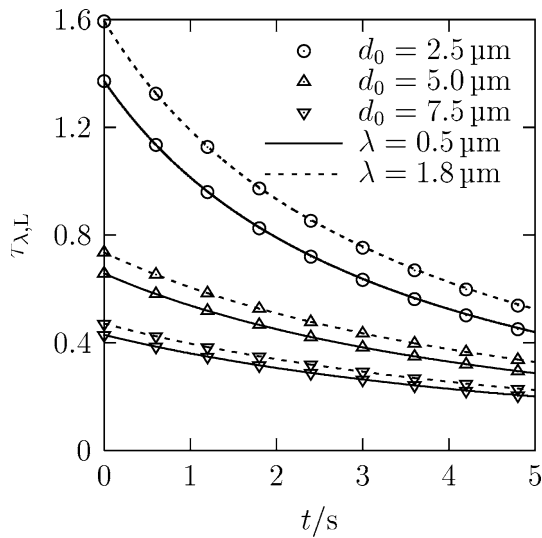


Fig. 8 Spectral optical thickness of the whole layer of coal particles undergoing steam-gasification, for radiation at $\lambda = 0.5$ and $1.8 \mu\text{m}$, and for initial particle diameters $d_0 = 2.5, 5$, and $7.5 \mu\text{m}$

4 Summary

Transient radiative heat transfer in a chemical reactive media was solved for a suspension of coal particles directly exposed to concentrated solar radiation and undergoing endothermic steam-gasification. The coal particles underwent shrinking at a rate determined by the reaction kinetics. The transient variation of the radiative properties (absorption and scattering coefficients and scattering phase function) of the shrinking particles, as well as their spectral dependence, was accounted for in the calculation of the temperature profiles, reaction extent, and overall radiative transfer. Parallel filtered collision-based MC method was applied to compute the radiative heat flux. The finite

volume method with explicit Euler time integration scheme was applied for solving the energy equation. In general, this model can be applied for solving transient radiative heat transfer problems involving high-temperature solid-gas thermochemical reactions with changing particle size, as it is typical for combustion, decomposition, cracking, and gasification processes. The model can be further extended to include poly dispersions, non-isothermal particles, and gas radiation, and its coupling to heat transport phenomena other than radiation.

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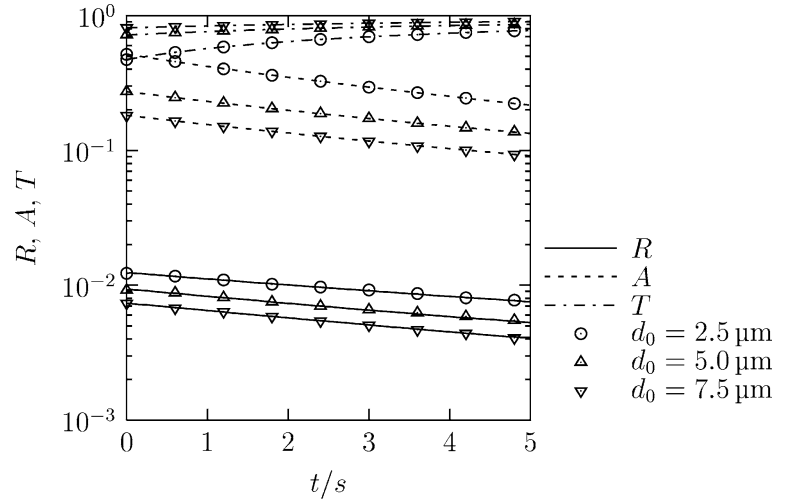
5 Appendix

5.1 Accuracy of the combined Monte Carlo method with time integration schemes

The accuracy of MC combined with the different time integration schemes is determined by solving for the temperature distribution within a plane layer of a gray-isotropic participating medium, and comparing the results with those obtained using a semi-analytical method. A 1D plane layer of a non-isothermal absorbing, emitting, and isotropically scattering gray medium is considered. The medium is contained within black plane boundaries at constant temperature T_b . Its initial temperature is T_0 . The equations developed are valid for either $T_0 > T_b$ or $T_0 < T_b$, i.e. the medium undergoing either cooling or heating, respectively. Medium properties are listed in Table 3 and are assumed arbitrary. Neglecting convection, conduction, pressure work, and internal heat generation, the general energy equation for one-component medium is given by:

$$\rho c_v \frac{\partial T}{\partial t} = -\nabla \cdot \vec{q}_r'' \quad (24)$$

Fig. 9 Overall reflectance, absorptance and transmittance of the layer as a function of time for particles of initially $d_0 = 2.5, 5$, and $7.5 \mu\text{m}$



where \bar{q}_r'' is the radiative heat flux across the medium. Two approaches for finding the divergence of \bar{q}_r'' are presented: (1) the semi-analytical method, and (2) the MC method. Finite volume technique is employed in both approaches for discretization of Eq. 24. In particular, for a 1D geometry divided into a large number of sub-layers, each of same thickness and at uniform temperature:

$$T_j^{n+1} = T_j^n + \int_{\Delta t} \frac{1}{\rho_j c_{vj} \Delta x_j} \int_{\Delta x} \left(-\frac{\partial q_r''}{\partial x} \right) dx dt \quad (25)$$

where j denotes a sub-layer and $\Delta t = t^{n+1} - t^n$ is the time step interval.

5.1.1 The semi-analytical method

The radiative flux divergence can be analytically derived from the equation of radiative transfer [25]:

$$\frac{dq_r''}{dx} = 4 \frac{k}{\omega} [\sigma T^4(x, t) - \pi S(x, t)] \quad (26)$$

The source function can be expressed in terms of the exponential integral functions,

$$S(x, t) = (1 - \omega) \frac{\sigma T^4(x, t)}{\pi} + \frac{\omega}{2} \times \left\{ \frac{\sigma T_b^4}{\pi} [E_2(x) + E_2(L - x)] + \int_0^L S(x^*, t) E_1(|x^* - x|) dx^* \right\} \quad (27)$$

Inserting (26) into (25) and further assuming that the flux divergence and field variables are constant over each sub-layer, an expression for the temperature in each sub-layer can be derived:

$$T^{n+1} = T^n + \frac{4}{\rho c_v \omega} \frac{k}{\omega} \int_{t_n}^{t_{n+1}} [\pi S(x, t) - \sigma T^4(x, t)] dt \quad (28)$$

where the shorthand notation $T^n = T(t)^n$ is introduced. Note that for simplicity, spatial discretization in sub-layers is omitted from the notation of Eq. 28 and succeeding equations. The temperature distribution is obtained using Eq. 28, where the integral is computed using the different time integration schemes presented below.

5.1.2 The MC method

The integral of the divergence of the radiative flux over a sub-layer, shown in the right-hand side of Eq. 25, represents the net radiative power absorbed by that sub-layer, i.e. the difference between absorbed and emitted power:

$$q_r = q_a - q_e = \int_{\Delta x} \left(-\frac{\partial q_r''}{\partial x} \right) dx \quad (29)$$

Inserting Eq. 29 into Eq. 25 yields,

$$T^{n+1} = T^n + \frac{1}{\rho c_v \Delta x} \int_{t^n}^{t^{n+1}} (q_a - q_e) dt \quad (30)$$

Table 3 Baseline parameters used for the gray-isotropic medium

Parameter	Value	Unit
c_v	720	$\text{J} \cdot \text{kg}^{-1} \cdot \text{K}^{-1}$
f_v	10^{-5}	—
L	0.1	m
$L/\Delta x$	250	—
T_0	1000	K
T_b	300	K
ε_b	1	—
κ	10.7	m^{-1}
σ_s	9.3	m^{-1}
ρc	2250	$\text{kg} \cdot \text{m}^{-3}$

where

$$q_e = 4kV\sigma T^4 \quad (31)$$

and q_a is found by MC as the number of rays absorbed in a sub-layer times the power carried by a single ray. At the boundaries, the emitted heat fluxes are:

$$q''_{e,b_1} = -q''_{e,b_2} = \epsilon_b \sigma T_b^4 \quad (32)$$

Temperatures and, consequently, the total power emitted vary with time. In contrast, the total number of rays for each MC iteration is set constant with time. Thus, the power carried by each ray is calculated at each MC iteration as the ratio of the total power emitted to the total number of rays. The temperature distribution is obtained using Eq. 30, where the integral is computed using the different time integration schemes presented below.

5.1.3 Time integration schemes

Five schemes for time integration are considered [7, 5]:

$$\text{Explicit Euler : } T^{n+1} = T^n + f(t^n, T^n)\Delta t \quad (33)$$

$$\text{Implicit Euler : } T^{n+1} = T^n + f(t^{n+1}, T^{n+1})\Delta t \quad (34)$$

$$\begin{aligned} \text{Crank Nicolson: } T^{n+1} = T^n + [f(t^n, T^n) \\ + f(t^{n+1}, T^{n+1})] \frac{\Delta t}{2} \end{aligned} \quad (35)$$

$$\begin{aligned} \text{2nd - order Runge - Kutta: } T^{n+\frac{1}{2}} = T^n + f(t^n, T^n) \frac{\Delta t}{2} \\ (36) \end{aligned}$$

$$T^{n+1} = T^n + f\left(t^{n+\frac{1}{2}}, T^{n+\frac{1}{2}}\right)\Delta t \quad (37)$$

$$\begin{aligned} \text{4th - order Runge - Kutta: } T^{n+\frac{1}{2},a} = T^n + f(t^n, T^n) \frac{\Delta t}{2} \\ (38) \end{aligned}$$

$$T^{n+\frac{1}{2},b} = T^n + f\left(t^{n+\frac{1}{2}}, T^{n+\frac{1}{2},a}\right) \frac{\Delta t}{2} \quad (39)$$

$$T^{n+1,a} = T^n + f\left(t^{n+\frac{1}{2}}, T^{n+\frac{1}{2},b}\right)\Delta t \quad (40)$$

$$\begin{aligned} T^{n+1} = T^n + [f(t^n, T^n) + 2f\left(t^{n+\frac{1}{2}}, T^{n+\frac{1}{2},a}\right) \\ + 2f\left(t^{n+\frac{1}{2}}, T^{n+\frac{1}{2},b}\right) + f(t^{n+1}, T^{n+1,a})] \frac{\Delta t}{6} \end{aligned} \quad (41)$$

where f denotes a time-dependent function. The explicit Euler and both Runge-Kutta schemes are self-starting. They only require an initial temperature distribution at $t=0$. The implicit Euler and Crank-Nicolson schemes require, in addition, the temperature distribution at time level $n+1$, which is found iteratively. The MC method requires ray-tracing iterative runs at each time level, i.e.

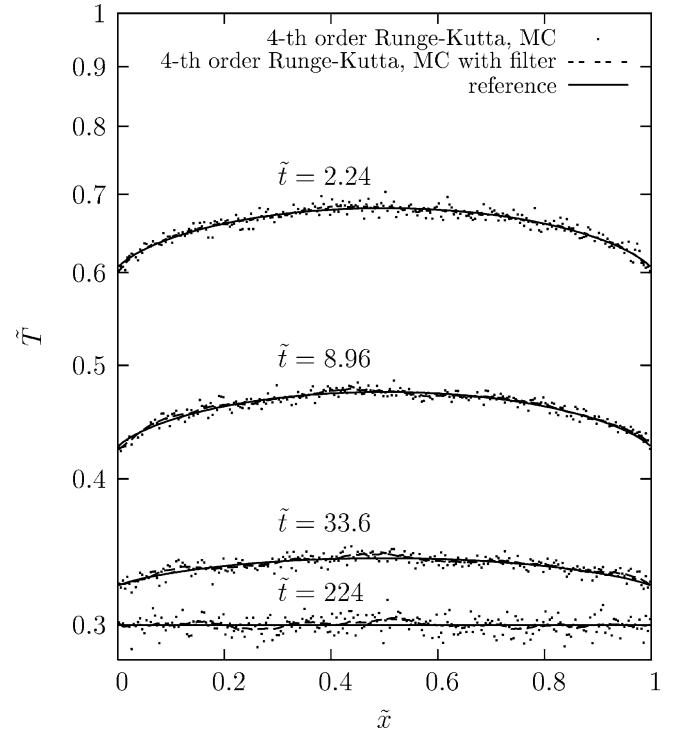


Fig. 10 Temperature profiles calculated by the “reference” semi-analytical method and by the filtered and unfiltered MC method using $n_{\text{rays}} = 10^5$ and the 4th-order Runge-Kutta time integration scheme

each time the integral on the right-hand side of (30) is evaluated. As long as such a procedure is followed, the generalization of the MC method to problems involving time-dependent radiation properties is straightforward and only requires the additional computation of these properties at each time level.

5.1.3.1 Parallelization and smoothing filters for MC

The described MC procedure requires significant computational resources when applied in conjunction with time integration schemes. Parallelization and smoothing filter concepts help alleviate this problem. Parallel methods are especially suitable for MC because they take advantage of the fact that tracing a generic ray constitutes a Markov chain's event and, consequently, can be treated fully independent of tracing other generic rays. Different concepts of parallelization of MC are outlined in [4]. In this work we use parallelization by spatial region for the energy balance and by ray for ray tracing. Parallel MC computing concept has been previously applied for steady-state radiation problems involving iterative ray tracing [13], and it is here applied for the first time to transient radiation heat transfer. The proposed mechanism encompasses a master process and several slave processes, each responsible for tracing assigned rays within the whole domain. Material properties and grid configuration are obtained at the master process and forwarded to all slave processes, where the temperatures of the assigned sub-layers are computed by

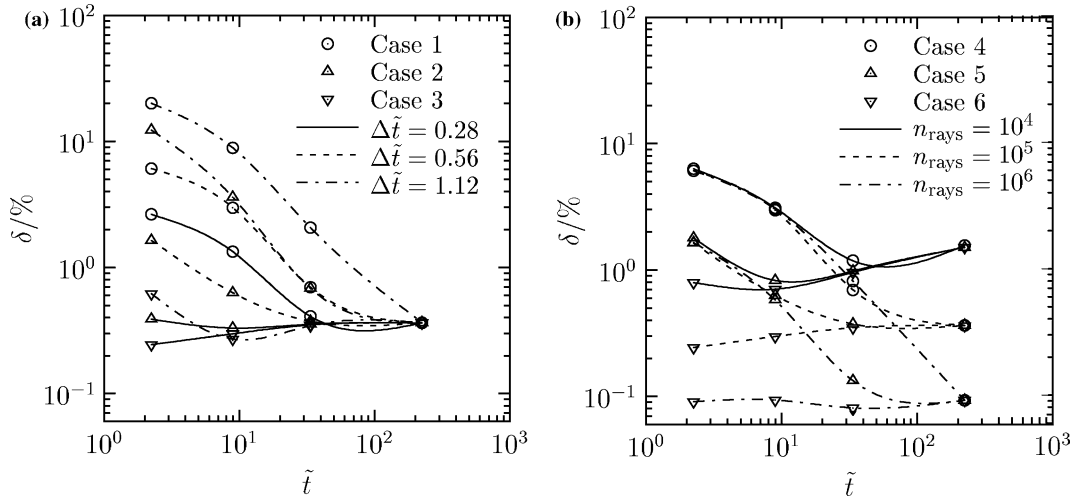


Fig. 11 Error of the MC method (cases are listed in Table 4)

summing up q_a obtained from all processes. Subsequently, the time integration schemes are used to move forward in time. Critical to the implementation of this mechanism is the correct generation of pseudo random numbers and this study uses the algorithm developed by Press et al. [22]. The spatial noise level in the MC solution was reduced by employing a 33-point quadratic lower-pass Savitzky–Golay smoothing filter [22].

5.1.4 Results

The MC was performed using all of the aforementioned time schemes, for samples containing $n_{\text{rays}} = 10^4, 10^5$, and 10^6 rays, for time steps $\Delta t = 2, 4$, and 8×10^{-3} s with and without smoothing filter. In addition, the semi-analytical method was applied using the 4th-order Runge-Kutta time integration scheme for $\Delta t = 10^{-3}$ s, yielding the most accurate solution, referred to as the “reference” solution. Common baseline parameters are listed in Table 3. Since for this specific example material properties are assumed constant with time, results are presented in a non-dimensional form:

$$\tilde{t} = \frac{4\sigma T_0^3}{f\sqrt{\rho c_v L}} t \quad (42)$$

$$\tilde{T} = \frac{T}{T_0} \quad (43)$$

Figure 10 shows the temperature profiles for the reference solution and for the filtered/unfiltered MC results using $n_{\text{rays}} = 10^5$ with the 4th-order Runge-Kutta integration scheme. The unfiltered results are indicated by the dots scattered around the reference solution. Scattering increases as steady state is approached. The temperature profile for the filtered results coincides well with the reference curve. The accuracy of the MC method is determined by calculating the error in the temperature profile based on the lumped squared relative differences between the actual and the reference solution, according to:

$$\delta^2(t) = \frac{1}{L} \int_L \left[1 - \frac{T(x,t)}{T_{\text{ref}}(x,t)} \right]^2 dx \quad (44)$$

where T_{ref} denotes the temperature distribution obtained for the reference (semi-analytical) solution and T the one obtained for the MC solution. This error is caused by the approximation in the time and space discretization due to a finite time and space intervals, and by the statistical approximation due to a finite sample of rays. Computations with refined space grid were carried out until the error due to space discretization could be neglected for a space grid resolution $L/\Delta x = 250$. Figure 11 shows the error in the filtered MC solution, calculated by Eq. 44, as a function of time for the cases listed in Table 4.

Obviously, the accuracy is improved with shorter time step intervals, larger samples of rays, and with the use of a filter. The explicit Euler scheme exhibited the

Table 4 Parameters for the MC method

Case	n_{rays}	$\Delta \tilde{t}^a$	Time-integration scheme
1	10^5	0.28, 0.56, 1.12	Explicit Euler
2	10^5	0.28, 0.56, 1.12	2nd-order Runge-Kutta
3	10^5	0.28, 0.56, 1.12	4 th -order Runge-Kutta
4	$10^4, 10^5, 10^6$	0.56	Explicit Euler
5	$10^4, 10^5, 10^6$	0.56	2nd-order Runge-Kutta
6	$10^4, 10^5, 10^6$	0.56	4 th -order Runge-Kutta

^a Non-dimensional time step $\Delta \tilde{t} = 0.28, 0.56$ and 1.12 , which for the properties of Table 3 correspond to $\Delta t = 2, 4$, and 8×10^{-3} s, respectively

best stability with longer time steps and minimum computational time. As expected, the 4th-order Runge-Kutta scheme gives the highest accuracy, with an error of less than 1% for $n_{\text{rays}} \geq 10^5$ and $\Delta t \leq 1.12$. Increasing the sample of rays by an order of magnitude decreases the error roughly by half an order of magnitude. The introduction of filters further improves the accuracy such that the error for $n_{\text{rays}} = 10^5$ and without filter is comparable to the one for $n_{\text{rays}} = 10^4$ and with filter. For the implicit Euler or the Crank-Nicolson schemes, the MC method does not converge. For the 2nd-order or the 4th-order Runge-Kutta schemes with $\Delta t \geq 1.12$, $n_{\text{rays}} \leq 10^4$, and without low-pass filter, the MC method does not converge either.

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